## PDF and NMR Study of Ordering in the Positive Electrode Material $Li(NiMn)_{0.5}O_2$

Julien Bréger<sup>a</sup>, Nicolas Dupré<sup>a</sup>, Peter J. Chupas<sup>b</sup>, Peter L. Lee<sup>b</sup>, Thomas Proffen<sup>c</sup>, John B. Parise<sup>a</sup>, Clare P. Grey<sup>a</sup>, <sup>a</sup>Department of Chemistry, SUNY Stony Brook, USA. <sup>b</sup>Advanced Photon Source, ANL, USA. <sup>c</sup>Los Alamos Neutron Science Center, LANL, USA. E-mail: jbreger@ic.sunysb.edu

The local environments and short-range ordering of  $Li(NiMn)_{0.5}O_2$ , a potential Li-ion battery positive electrode material<sup>1,2</sup>, were investigated by using a combination of X-ray and neutron diffraction and isotopic substitution, Li MAS NMR spectroscopy and, for the first time, X-ray and neutron Pair Distribution Function (PDF) analysis, associated with Reverse Monte Carlo (RMC) calculations.  $Li(NiMn)_{0.5}O_2$  adopts the  $LiCoO_2$  structure and comprises separate Li layers, transition metal (Ni,Mn) layers and O layers.

NMR experiments and Rietveld refinements showed that there is 10% of Li/Ni site exchange. Neutron PDF analysis revealed considerable local distortions in the layers that are not captured in the LiCoO<sub>2</sub> model. Large clusters were built to investigate cation ordering, by performing RMC calculations. Both NMR and RMC were consistent with a non-random distribution of Ni, Mn and Li cations in the transition metal layers. Constraints from both methods showed the presence of short-range order in the transition metal layers comprising LiMn<sub>6</sub> and LiMn<sub>5</sub>Ni clusters combined with Ni and Mn contacts that are consistent with those found in some of the proposed structures based on Li<sub>2</sub>MnO<sub>3</sub>-like ordering of the cations.

[1] Ohzuku T., Makimura Y., Chem. Lett., 2001, 744. [2] Lu Z., MacNeil D. D., Dahn J.R., Electrochem. Solid-State Lett., 2001, 4, A191.

Keywords: lithium batteries, pair distribution function, NMR