

Intermolecular interactions in 1,1'-Binaphthyl, polymorphs and symmetry breaking

C. Ignacio Sainz-Díaz^a, Africa P. Martín-Islán^b, Julian H. E. Cartwright^a, Alfonso Hernández-Laguna^b, ^a*Instituto Andaluz de Ciencias de la Tierra, CSIC - Universidad de Granada, Av. Fuentenueva s/n, 18002-Granada (Spain). sainz@lec.ugr.es.*
^b*Estación Experimental del Zaidín, CSIC, C/ Profesor Albareda, 1, 18008-Granada (Spain). E-mail: sainz@lec.ugr.es*

A theoretical-experimental work is presented related with the chiral symmetry breaking of melting crystallization of 1,1'-binaphthyl derivatives and polymorphism. We confirm that the chiral symmetry breaking can be observed in crystallization from a melt of 1,1'-binaphthyl by a constant stirring during the crystallization. Crystallographic studies by Powder X-ray diffraction (PXRD) reveal two crystallographic forms of 1,1'-binaphthyl: one chiral form (P4₂1₂) with either R or S enantiomers of the trans-1,1'-binaphthyl conformer and another racemic crystal (C2/c) with both enantiomers of the cis-1,1'-binaphthyl conformer. Quantum mechanical calculations of the crystal lattice for 1,1'-binaphthyl and 2,2'-dihydroxy-1,1'-binaphthyl polymorphs were performed by Density Functional Theory approximation. Our calculations reproduce the crystal lattice parameters and PXRD pattern finding the P4₂1₂ form with lower energy than the C2/c form for 1,1'-binaphthyl. The main intermolecular interactions in 1,1'-binaphthyl crystals are weak aromatic CH/ π hydrogen bonds, which are responsible for enantiomeric discrimination in the molecular recognition during crystallization. The C2/c form achieves a more efficient packing than the chiral one, but intermolecular interactions in P4₂1₂ form are stronger than in C2/c form. In 2,2'-dihydroxy-1,1'-binaphthyl the intermolecular interactions are stronger with hydrogen bonds between the hydroxyl groups and polymorphs can be predicted by Monte Carlo simulated annealing.

Keywords: 1,1'-binaphthyl, intermolecular interactions, chiral recognition