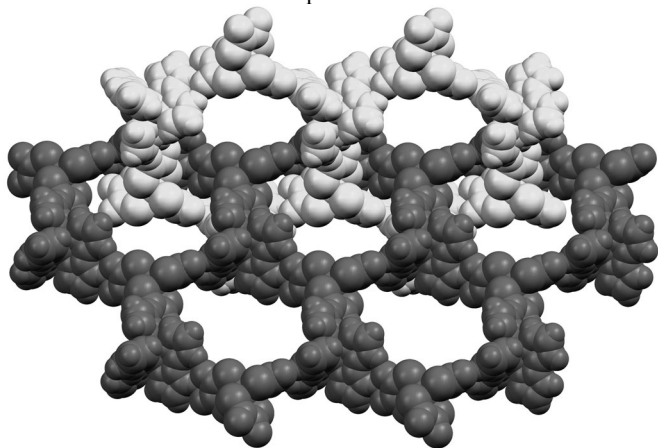


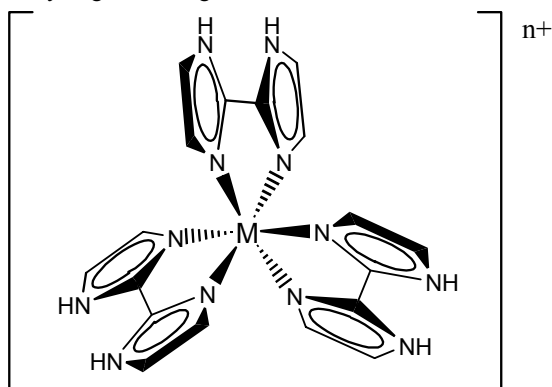
### Synthesis of 2D and 3D Nets using Biimidazole Complexes

Alexandra M. E. Griffin, A. Guy Orpen, *School of Chemistry, University of Bristol, Bristol BS8 1TS*. E-mail: a.griffin@bristol.ac.uk

The use of hydrogen-bonding to control the arrangements of cations and anions has been well documented in previous research.[1] [2] Here we report tris-chelated metal complexes containing the biimidazole ligand, which are capable of hydrogen-bonding to form three-dimensional and interpenetrated two-dimensional nets.



In particular  $[\text{Ni}(\text{H}_2\text{biim})_3]^{2+}$  and  $[\text{Cr}(\text{H}_2\text{biim})_3]^{3+}$  ( $\text{H}_2\text{biim}$  = 2,2'-biimidazole) have been crystallized with a variety of anionic tectons capable of hydrogen bonding to these cations.



[1] Podesta T. J., Orpen A. G., *CrystEngComm.*, 2002, 4, 336. [2] Angeloni A., Orpen A. G., *Chem. Commun.*, 2001, 4, 343.

**Keywords:** crystal synthesis, hydrogen bonding, biimidazole complexes