## A Lower Rim Tert-butyl Calix[4]arene Derivative that Binds Ag(I)

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The X-ray structures of a tert-butyl calix[4]arene derivative with -O-H and -O-(CH<sub>2</sub>)<sub>2</sub>-NH-(P=S)(OCH<sub>2</sub>CH<sub>3</sub>)<sub>2</sub> pendant arms (EPS) as molecular complexes with methanol, EPS(MeOH), 1, with dichloromethane, EPS(Cl<sub>2</sub>CH<sub>2</sub>), 2, and its complex with silver(I) and acetonitrile, AgEPS(MeCN)K.(ClO<sub>4</sub>)<sub>2</sub>.H<sub>2</sub>O, 3, were determined. The structures were solved from 17027 (1), 8810 (2) and 7729 (3) reflections with  $I \ge 2\sigma(I)$  collected at T=100 K, and refined to R1factors of 0.054 (1), 0.049 (2) and 0.090 (3). In all compounds, a pair of strong O-H...O(pend) bonds in the lower rim produces relatively open chalices. This, in turn, promotes the hydrophobic interaction with the solvent molecules found included in these cavities. The upper half of the EPS pendant arms is anchored by N-H...O(ox) intramolecular bonds where now the phenol oxydryl oxygen atoms act as H-bond acceptors. Despite the relative rigidity of the upper half of the hydrophilic cavity, the lower half is pre-organized towards soft metal ion complexation at the thiophosphorous sulphur binding site by unhindered torsion movements, mainly around the N-P σ-bond. The silver(I) complex with **EPS**, exhibits an approximate  $C_2$  symmetry around the calix axis. The Ag(I) ion is sited on this axis, linearly coordinated to the sulphur atoms of the pendant arms [Ag-S distances of 2.414(2) and 2.422(2) Å,  $\angle$ (S-Ag-S)=171.99(8)°]. We thank FAPESP (Brazil) and EU (Contract ICA CT 2000-30006) for support. Keywords: calixarene complexes, environmental chemistry, supramacromolecules