

## Structural Study of Sn-doped In<sub>2</sub>O<sub>3</sub>

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Tin doped In<sub>2</sub>O<sub>3</sub> (ITO) is widely used in phototronic, optoelectronic and microelectronic devices. There are numerous studies on ITO, but the understanding of its defect structure is rather incomplete. In<sub>2</sub>O<sub>3</sub> possesses a cubic structure, the space group Ia<sup>3</sup>[1]. The structure contains two different six-fold-coordinated cation sites referred as sites B and D, respectively. This paper focuses on X-ray diffraction and <sup>119</sup>Sn Moessbauer studies of the polycrystalline ITO samples containing 2-14 at% Sn. Nanocrystalline ITO powders were prepared by a sol-gel technique using InCl<sub>3</sub> and SnCl<sub>4</sub> reagent grade chemicals and annealed at 300°C for 5 h. Lattice parameter *a* of doped In<sub>2</sub>O<sub>3</sub> increased with Sn-doping level up to 8 at% and decreased above. Such behavior of lattice parameter suggests that tin substitution for In<sup>3+</sup> on sites B and D is non-uniform and depends on tin content. <sup>119</sup>Sn Moessbauer spectra indicated that only the Sn<sup>4+</sup> state is present in ITO samples. The least square fitting of spectra was performed by assuming presence of two doublets. In accordance with [2], doublets with isomer shifts (IS) in the range 0.09–0.17 and 0.36–0.464 mms<sup>-1</sup>, respectively, correspond to the D and B sites in the cation sublattice of the In<sub>2</sub>O<sub>3</sub> structure. The area ratio of the two doublets depended on tin content.

[1] Marezio M., *Acta Cryst.*, 1966, **20**, 723. [2] Binczycka H., et al., *Phys. Stat. Sol. (B)*, 2005, *in press*.

**Keywords:** Sn-doped In<sub>2</sub>O<sub>3</sub>, x-ray diffraction, Moessbauer spectroscopy