

## Non Classical vs. Classical Metal...H<sub>3</sub>C-C Interactions: A Neutron Diffraction Study of a 14-Electron Ruthenium(II) System

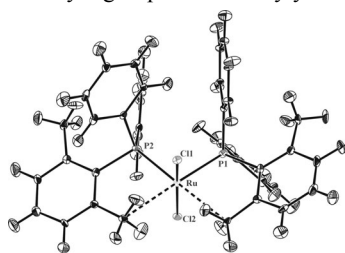
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A neutron diffraction study establishes the precise nature of the  $\delta$  agostic interactions in the complex RuCl<sub>2</sub>[PPh<sub>2</sub>(2,6-Me<sub>2</sub>C<sub>6</sub>H<sub>3</sub>)]<sub>2</sub> (**1**). By contrast to the classical agostic bonding, it is shown that two ortho-methyl groups of the xylyl substituents interact with the



unsaturated metal centre through two C-H bonds each. The result is also substantiated by the NMR data in solution. [1], [2] Re-examination of all the X-ray structures with  $\beta$ ,  $\gamma$ ,  $\delta$  and  $\epsilon$  M...H<sub>3</sub>C-C moieties as well as DFT calculations on models of **1** allow to

conclude that the agostic interactions span the range between the classical (M... $\eta^2$ -HC) and the non-classical (M... $\eta^3$ -H<sub>2</sub>C) types, depending on the number of atoms between the metal and the methyl group.

[1] Baratta W., Mealli C., Herdtweck E., Ienco A., Mason S. A., Rigo P., *J. Am. Chem. Soc.*, 2004, **126**, 5549. [2] Baratta W., Herdtweck E., Rigo P., *Angew. Chem. Int. Ed.*, 1999, **38**, 1629.

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